

Dugway Proving Ground



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Inactivation of *Bacillus anthracis* Spores in Drinking Water by Mixed Oxidant Solution

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Background

- Soldiers in some tactical scenarios do not have access to quartermaster potable water supplies.
- When calcium hypochlorite is used as a water disinfectant, a high free available chlorine concentration (FAC) is required to kill *B. anthracis* spores.
- Hypochlorite is toxic at levels around 190 mg/kg body weight.



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Background, cont.

- Water treated with calcium hypochlorite at a level to kill anthrax spores is neither potable nor palatable.



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Chemistry

- FAC is the concentration of $\text{Cl}_2 + \text{HOCl} + \text{OCl}^-$ (expressed as Cl_2) and is the most chemically reactive and biocidal form of chlorine commonly used for disinfection
- Hypochlorous acid (HOCl) is the predominant form of FAC at $\text{pH} < 7.2$.
- Optimum biocidal effectiveness of FAC occurs at $\text{pH} 6.0$ to 7.5 .



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Chemistry, cont.

- At high pH HOCl dissociates primarily to hypochlorite ion ($\text{OCl}^- + \text{H}^+$).
- OCl^- is about 1/100 as effective a biocide as is HOCl, but still effective.
- HOCl at $\text{pH} < 4$ forms chlorine gas (Cl_2).
- Combined chlorines are formed from the reaction of FAC with nitrogenous compounds such as ammonia and urea.



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Chemistry, cont.

- Reactions with ammonia or amino group-containing organic compounds produce chloramines and organic chloramines.
- Chloramines are formed in a step-wise manner as FAC is added resulting in
 - Monochloramine (NH_2Cl)
 - Dichloramine (NHCl_2)
 - Trichloramine (NCl_3)



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Chemistry, cont.

- Chloramines are weak oxidizing agents and biocides relative to both HOCl and OCl⁻.
- One source suggests that “monochloramine may only have 5% of the disinfecting power of FAC on a time/concentration basis.”



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Mixed Oxidants

- Produced by the electrolysis of sodium chloride solution
- Complete characterization of the molecular species present has not been possible.
- Characteristics of mixed oxidant activity suggest species are present in addition to HOCl and OCl⁻ (ie. FAC)



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Mixed Oxidant Advantages

- Rapid disinfection rates
- Removal of biofilms from surfaces of water distribution systems and treatment works
- More stable chlorine residual



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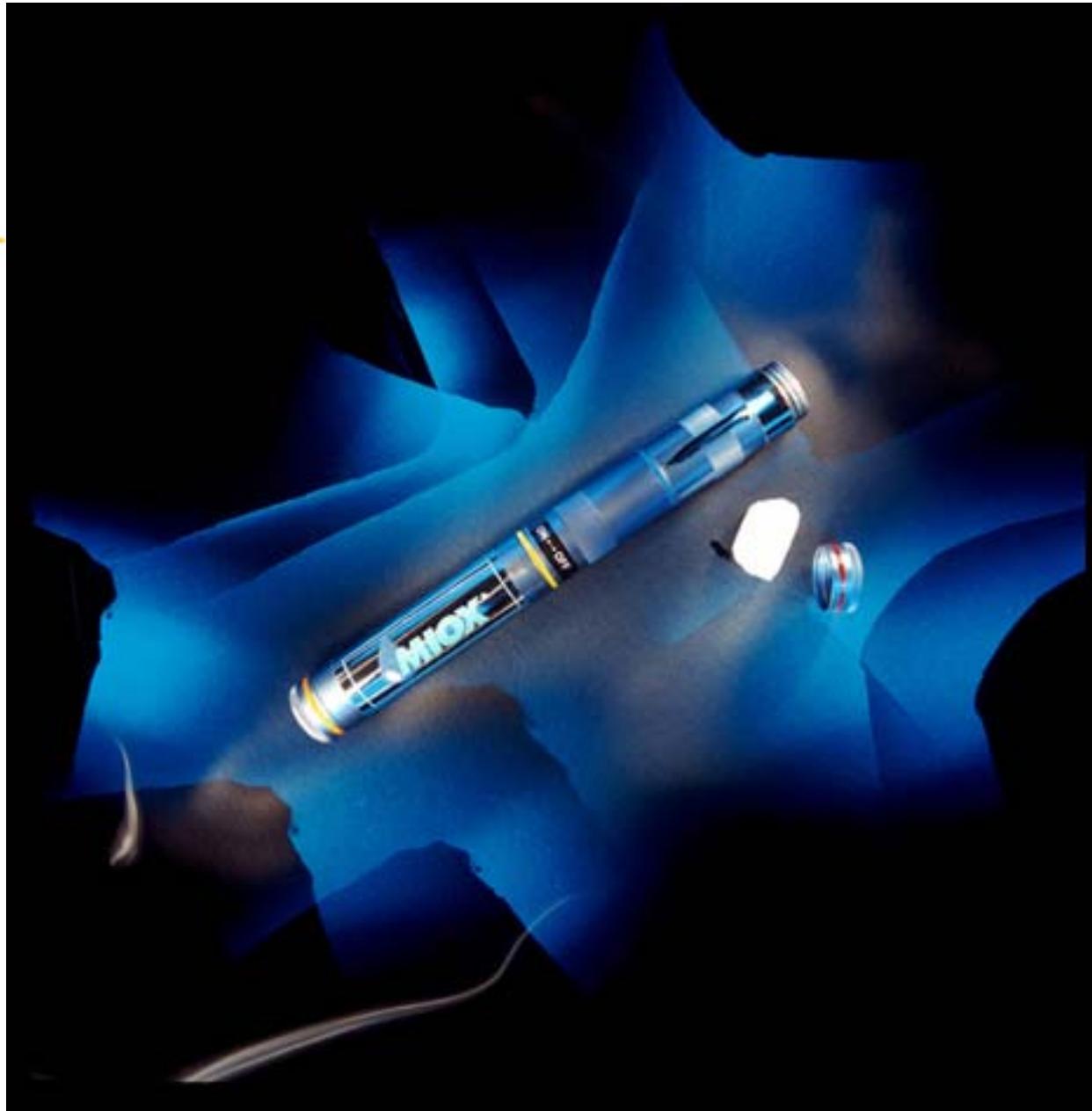
Disinfection Pen

- MIOX Corporation
- Los Alamos Technical Associates funded by DARPA to investigate technology for single soldier use
- Life Sciences Division tasked to provide independent evaluation using simulants and vaccine strains.



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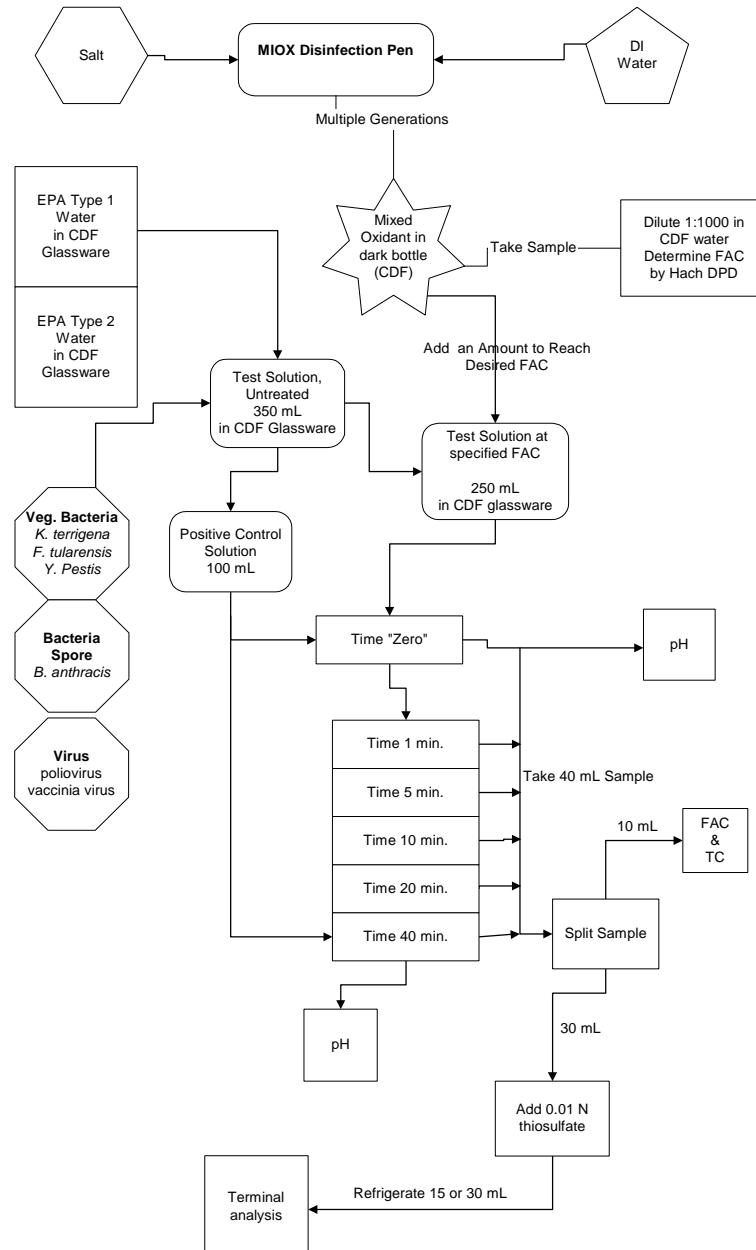
Initial Dugway Studies

- EPA Type 1 Water
 - Free of any chlorine or other disinfectant residue
 - pH 6.5 to 8.5
 - Total organic carbon 0.1 - 5.0 mg/L
 - Turbidity 0.1 to 5 NTU
 - Temperature $20\pm5^{\circ}\text{C}$
 - Total dissolved solids (sea salt) 50 - 100 mg/L



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Initial Studies, cont.

- *Bacillus anthracis* strain Sterne
 - spores
 - washed
- Looked for a 6 log reduction
- Challenge dose of 1E7 cfu/mL
- Details of test published as WDTC Document WDTC-IR-01-047



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Initial Studies, cont.

- Result: The 7 log cfu/mL challenge dose was killed in 40 minutes when treated with MOS at a FAC of 160 mg/L
- Analysis: It would take a soldier 40 MOS generation cycles of the pen to produce the required doses in a 1-liter canteen.
- Conclusion: Impractical



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Follow-on Studies

- *B. anthracis* vegetative cells known to be sensitive to chlorine inactivation
- Pre-treat water with a germinant solution
- Phil Hanna (Univ. of Michigan) suggested
 - 10 mM L-alanine
 - 1 mM serine
 - 1 mM inosine

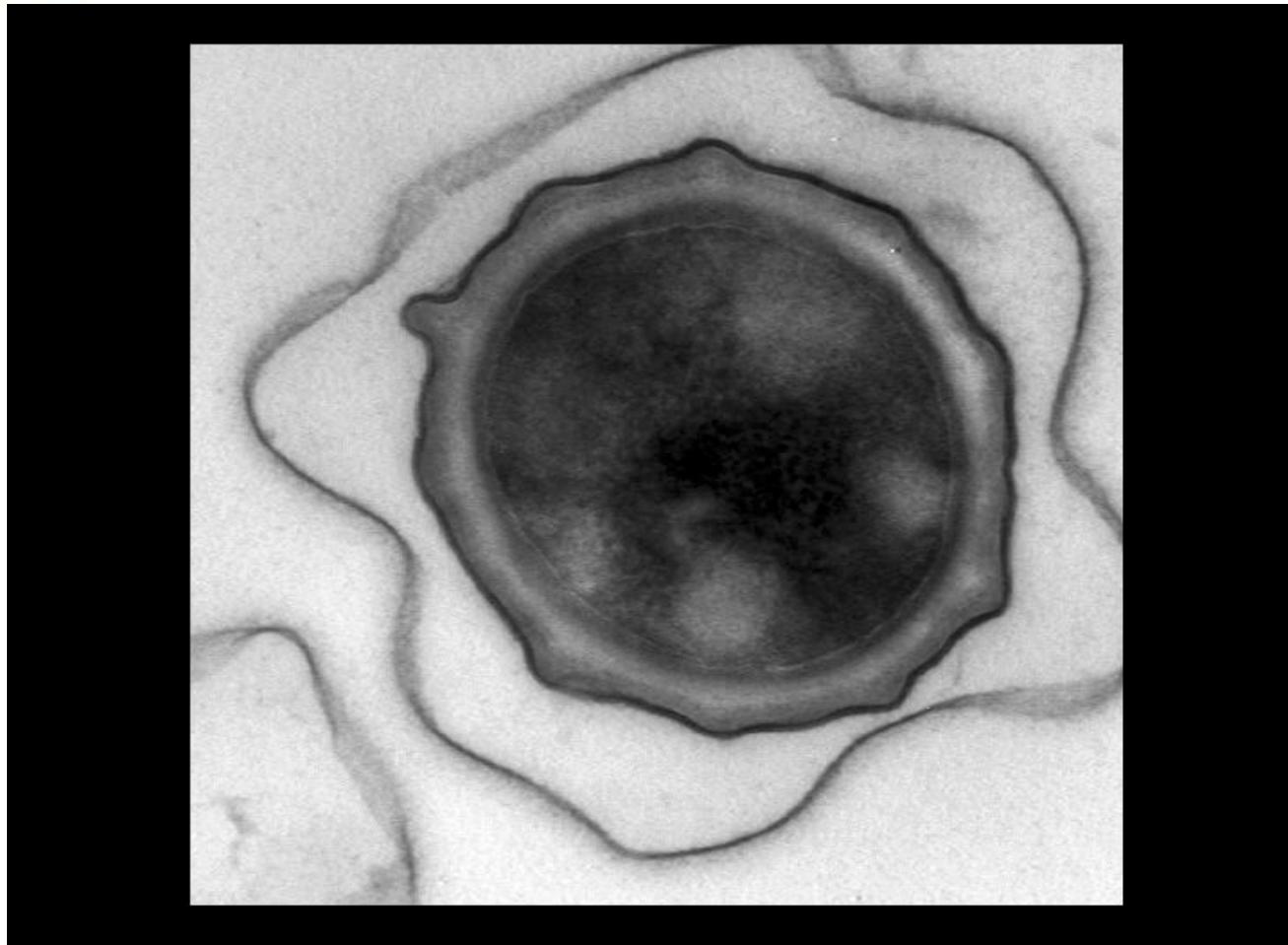


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Anthrax Spore Ultrastructure

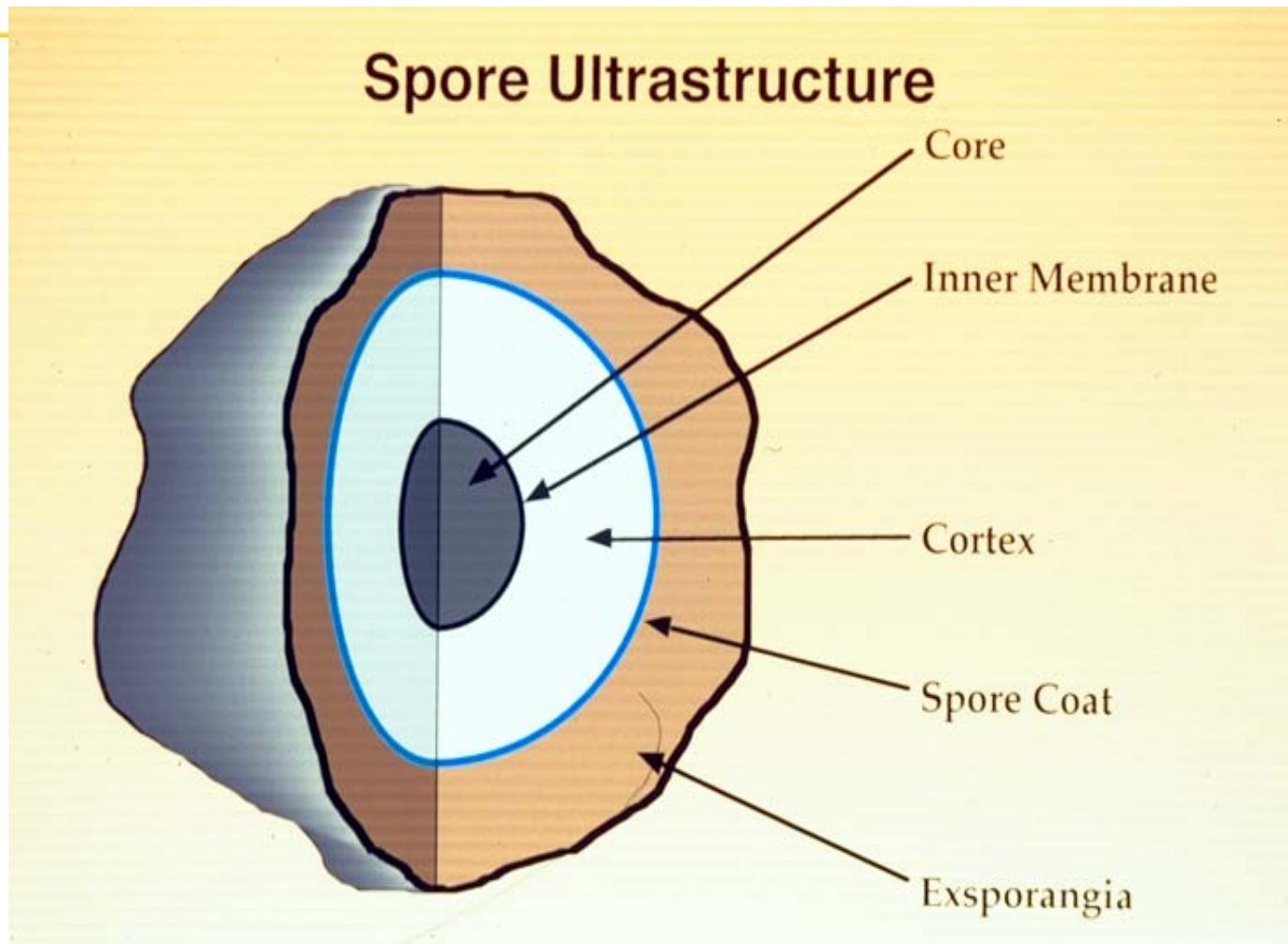


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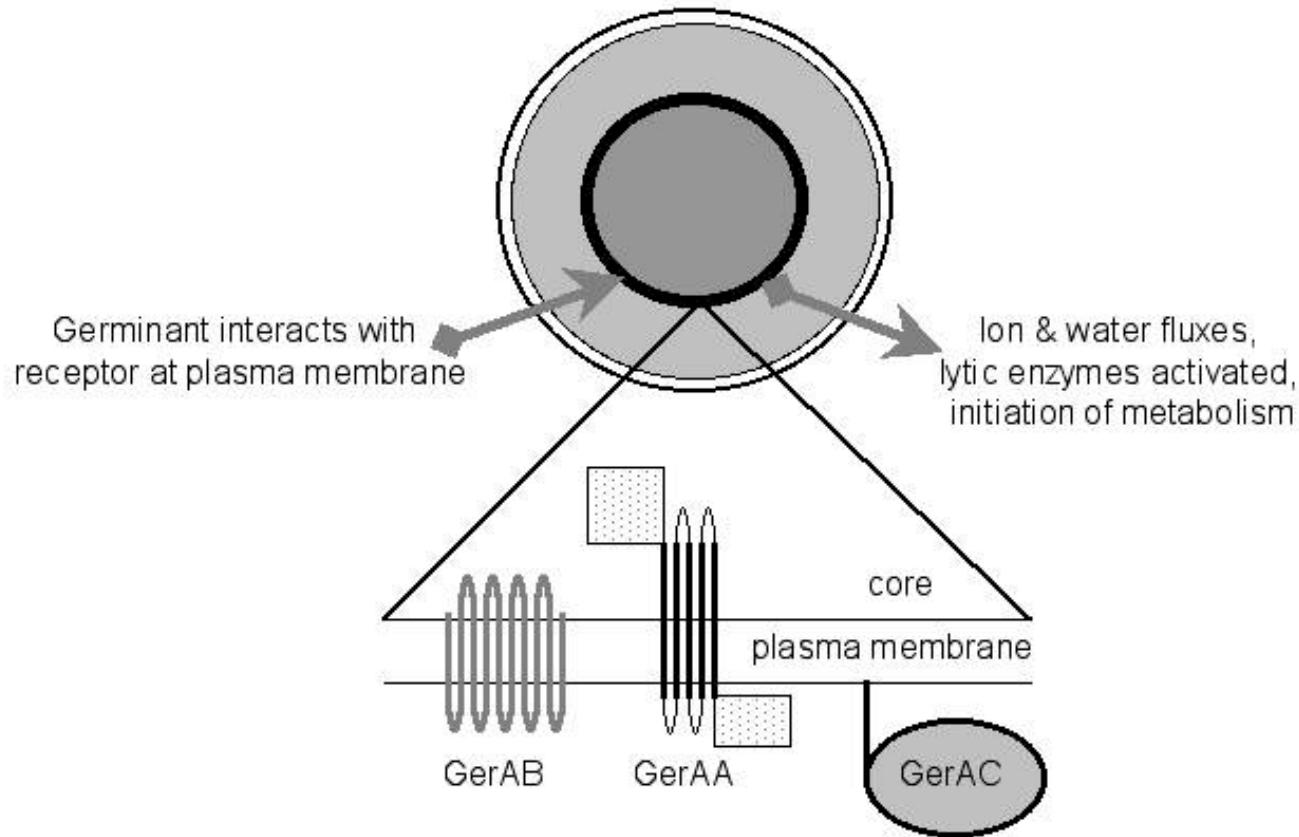


Anthrax Spore Ultrastructure



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Germinant Method A

- Used sterile tap water (no chlorine demand)
- Add Sterne spores to concentration of 1E6 cfu/mL
- Add germinant, incubate 30 minutes
- Add MOS to calculated FAC of 160 mg/L
- Plate samples at 1, 5, 10, 15, 25, 35, 45, & 60 minutes; incubate overnight



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Germinant + MOS Results

- Sterne spores would germinate using concentrations 1/10 of those recommended:
 - 1 mM L-Alanine
 - 0.1 mM Serine
 - 0.1 mM Inosine
- Our measure of germination being sensitivity to MOS.
- Germination results consistent with those previously published



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Method A Results, cont.

- Chlorine demand, germinant ~ 155 mg/L over 25 minutes
- Spores killed in 25 minutes after adding MOS at a final FAC decreasing to ~5.
- Significant increase in killing power of MOS on germinated spores
- Not unexpected



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Germinant Method B

- Same as Method A except that germinant and MOS were added to the spore suspension within less than 10 seconds of each other



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Method B Results

- Same as from Method A
- Spores killed in 20-30 minutes without a significant (<10 sec) incubation in germinant.
- **Unexpected result**
- No biochemical explanation evident
- Obvious that the majority of the FAC was being combined with germinants to form organic chloramines almost instantaneously.



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Organic Studies B

- The preceding results suggested another set of experiments.
- Looking at our results not as a function of germination but as a function of organic chloramine species and the N content of the germinants.



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Organic Chloramine Studies, cont.

- Theoretical framework from water treatment chemistry
 - A FAC ratio of $5 \times N$ mass concentration (ie. 5 mg FAC/mg N) + 25% should generate organic monochloramine with negligible amounts of organic di- and trichloramine
 - The 25% additional FAC satisfied non-N oxidant demand, determined separately.



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Organic Chloramine Studies, cont.

- A FAC ratio of $7 \times N$ mass concentration + 25% should generate a combination of organic chloramines, NOT primarily monochloramine
- A FAC ratio of $7.6 \times N + 25\%$ should drive the reaction to breakpoint where the N is released from the chloramines as N_2 and FAC is reduced to Cl^- (ie. both FAC and chloramines reduced in concentration in the water)



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Organic Chloramine Experiments

- Prepare germinant solution such that $N = 21$ mg/L in $1E6$ cfu/mL Sterne spore suspension
- Dose with 5 FAC $\times N$ or 7 FAC $\times N$
- React for 30 minutes at room temperature
- Dose with $7.6 \times N$ to cause breakpoint
- Plates samples at $1, 5, 15, 25, 40, 60,$ & 90 minutes (no thiosulfate added)



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L-Ala, Ser, & Inosine: 5 x N

Time (min)	FAC	TC	cfu/mL
1	40	79	5.9E3
5	21	51	1E2
15	14	30	0



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L-Ala, Ser, & Inosine: 7 x N

Time (min)	FAC	TC	cfu/mL
1	17	32	9E1
5	11	23	3E1
15	7	17	0



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Organic Chloramine Conclusions

- Both the 7 x N and 5 x N doses against the complete germinant kill the spores within 15 minutes.
- The 5 x N dose has a flatter reaction curve.
- No clear difference between a solution presumably consisting of monochloramines versus a chloramine cocktail.



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Inorganic Chloramine Study

- Substitute ammonium chloride as the nitrogen source because chloramine generation from this molecule simpler in concept.
- Believed this experiment would be a negative control: assumed organic germinants possess some inherent quality required to explain previous results



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Inorganic Chloramine Experiment

- Ammonium chloride (NH_4Cl) substituted for germinant solution at same nitrogen concentration (21 mg/L)
 - Preston & Douthit (J Gen Microbiol, 1984) described its co-germinant effects on *B. cereus*.
- A sample was also taken after 15 min treatment and before the second or “breakpoint” $7.6 \times N$ dose was added.



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NH₄Cl: 5 x N

Time (min)	FAC	TC	cfu/mL
1	9	47	0
5	8	22	0
15	7	17	0



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NH₄Cl: 7 x N

Time (min)	FAC	TC	cfu/mL
1	5	10	0
5	5	9	0
15	4.5	7.3	0



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15 Minute Samples

- $5 \times N = < 10 \text{ cfu/mL}$
- $7 \times N = 0 \text{ cfu/mL}$



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Inorganic Chloramine Conclusions

- A solution of chloramines derived from an inorganic source is capable of inactivating Sterne spores within 15 minutes.
- The effect was evident at 4 mg/L NH₄-N (down from 21 mg) but not at 2 mg/L suggesting a very pronounced activation threshold.
- Monochloramines are probably the most active species.



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Inorganic Conclusions, cont.

- At 4 mg/L NH₄-N, the MOS dose required (38 mg/L FAC) is practical using the latest version of the Disinfection Pen.
- Preston and Douthit (1983, 1988) conducted experiments with *B. cereus* in which the germinant activity of NH₄Cl was studied in the presence of Ala and other known co-germinants.
- Our results suggest that Sterne spores are sensitive to NH₄Cl alone, probably at some point downstream from the recognition points for organic co-germinants



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Inorganic Conclusions, cont.

- This effect has been confirmed in preliminary experiments at Phil Hanna's lab in the absence of MOS.



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